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Lignin and Cellulose Nanofibers from *Bambusa vulgaris* Schrad (Bamboo): An Extraction, Preparation and Characterization Study

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Abstract: In this paper, lignin and cellulose nanofibers were extracted and prepared from *Bambusa vulgaris* schrad (*B. vulgaris*: bamboo) before being subjected to characterization investigations. These extractions and preparations of the lignin and cellulose nanofibers were carried out chemically using alkali combined with bleaching treatments together with acid hydrolysis and sonication. The cellulose nanofibers were then subjected to morphological and dimensional characterization of the Zetasizer, Scanning Electron Microscopy (SEM) and Transmission Electron Microscopy (TEM) instruments. The functional groups investigation, using Fourier Transform infrared spectroscopy (FTIR), and thermal degradation via the Thermogravimetric analysis (TGA), of the bamboo lignin and of the cellulose nanofibers were also carried out. Results revealed that the percentage yields of the bamboo lignin and bamboo nanofiber were 21.91 wt% and 33.6 wt% respectively. The SEM and TEM investigations indicated the prepared nanofibers were rod-like in morphology, having sizes ranging from 20 to 100 nm. FTIR showed that the lignin extracted from bamboo typified G-S type lignin while the nanofibers are completely devoid of lignin. TGA revealed that the lignin was more thermally stable than the nanofiber under the test conditions. The obtained lignin and cellulose nanofibers showed promise for possible application as reinforcement agents in biodegradable nanocomposite film preparation.

Keywords: Bamboo, cellulose nanofibers, lignin, materials extraction, materials characterization.

1. Introduction

Cellulose and lignin are natural organic polymers found mainly in plants. Cellulose is the most widely available biopolymer having straight chain of D-glucose units, which are linked together via the bonds of $\theta(1 \rightarrow 4)$ -glycosides (Vazquez et al., 2013). It is a fundamental structural constituent of plants' primary cell wall, and of some species of algae, as well as of the oomycetes. It is also known that some species of bacteria also secrete cellulose for the formation of biofilms (Brethauer et al., 2020; Beloin et al., 2008). In contrast, lignin is a polymeric natural product resulting from trans-coniferyl alcohol, trans-sinapyl alcohol and trans pcoumaryl alcohol polymerization (Harman-Ware et al., 2017). Lignin is a complicated compound that is generally obtained from wood, while it is also a plants' secondary cell walls fundamental

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component, and that is also present in some algae (Börcsök et al., 2020; Labeeuw et al., 2015; Lebo, et al., 2015; Martone, et al., 2009). Since lignin and hemicellulose are covalently linked, lignin crosslinks diverse polysaccharides of plants, and by this, gives mechanical strength to the plant's cell wall, which extend as structural strength to the whole plant actually (Jawerth et al., 2020; Salmén et al., 2016; Gibson 2012; Chabannes et al., 2001).

Cellulose and lignin have been extracted from many plants using several methods that have been reported by many authors (Dinh Vu et al., 2017; Radotić & Mićić 2016; Yong et al., 2012; Kaushik et al., 2010; Alemdar & Sain, 2008). These extractions, of both cellulose and lignin, have been put into many applications, several of which have also been detailed in the literature (Gopakumar et al., 2018; Atanda, 2015; Zakikhani et al., 2014; Bao et al., 2011). For instance, fibers obtained from these extractions can be in micro or nano form. Currently, the nano form is favored owing to the exceptional effect such as the outstanding mechanical properties that could be obtained from only a little content of the nanofiller. A variety of methods have also been employed for preparing and extracting nanofibers of high purity from cellulosic materials (Shahi et al., 2020; Phanthong et al., 2018; Menon et al., 2017; Xiao et al., 2015; Lu et al., 2013; Saito et al., 2009; Paakko et al., 2007; Elazzouzi-Hafraoui et al., 2008; Alemdar & Sain 2008).

Bambusa vulgaris schrad (B. vulgaris: bamboo) is a naturally occurring biomass, which grows plentifully in most of the subtropical countries. Also, that B. vulgaris contains cellulose fibers imbedded in a lignin matrix makes it to be regarded as a

Received: January 24, 2023 Accepted: July 5, 2023 Published: March 31, 2025 composite material. This composite material exhibits many advantages, which among others includes strength, durability, light weight, stiffness, and biodegradability, and by these, *B. vulgaris* has found applications, from time memorial, in many sectors (Rahim et al., 2018; Atanda, 2015; Zakikhani et al., 2014; Atanda, 2015; Bao et al., 2011). Although *B. vulgaris* (bamboo) is readily available in Nigeria, it is highly underutilized (Atanda, 2015), and it has not been extensively studied as a source of nanofibers (Saniwan, et al., 2012; John et al., 2007; Chakraborty et al., 2006). Hence, the focus of this study is on the application of bamboo for preparing and characterizing nanofibers and lignin.

2. Materials and Methods

Materials

Sample materials and chemicals

Collections of the stems of *B. vulgaris* (bamboo) were from the banks of River Gurara at Izom, Niger State, Nigeria. These *B. vulgaris* stems were identified and authenticated at the herbarium of National Institute for Pharmaceutical Research and Development (NIPRD), Idu, Abuja, Nigeria, where a voucher of No NIPRD/H/6793 had been dropped for future reference. Purchased from reputable chemical stores for the study include analytical grades of C₂H₅COH, H₂SO₄ and H₂O₂ (from BDH Chemicals[®], England), NaOH (from Kermel[®], China), and HCI (from Griffin and George[®], England).

Equipment

The equipment employed in this study are: Oven (Gallenkamp[®] Size 2, 11526E), Autoclave (Patterson Scientific[®], Prestige 2100), Digital Weighing Balance (Ohaus Cooperation[®], China, Scout Pro SPU601), pH meter (Hanna[®], pH 212), Thermogravimetric analyzer (Perkin Elmer[®] STA 6000), Transmission Electron Microscope (Zeiss Auriga[®] HRTEM), Scanning Electron Microscope (Zeiss Auriga[®] HRSEM), Fourier Transform Infrared (Perkin Elmer[®], UK, Frontier FT-IR), Ultra Sonicator (Scientz[®], China, SB25-12DT), Zetasizer (Malven[®], USA, Nano-S series).

B. vulgaris Fiber and Lignin Preparation

The extraction of lignin from B. vulgaris followed the method that had been detailed in the research works reported by Yong et al. (2012), Kaushik et al. (2010) and Alemdar & Sain (2008). By these, stalks of B. vulgaris were sun-dried before being chopped to smaller pieces. This was followed by grinding the chopped pieces and subsequent screening to a mesh fraction of 40–60 μ m. From this ground stalk of B. vulgaris, 20 g was soaked in 4% w/w NaOH, for 24 h at room temperature, after which it was filtered and washed with distilled water until the complete elimination of the alkali. Further filtration was carried out for the second time before treatment with 10% w/w NaOH for 4 h in an autoclave maintained at 121 °C. Rewashing of the material with distilled water and subsequent filtration then followed. The filtrate was then acidified with H_2SO_4 to pH = 2 for the precipitation of lignin. The precipitated lignin were then filtered out of the mixture, separated and was then washed severally with water before being dried in an oven set at 40 °C. The supernatant ensuing from the

alkali treatment was bleached, at room temperature, in 8 % v/v H_2O_2 for 24 h to obtain *B. vulgaris* fiber, which was washed and filtered over and over again, just like before, for obtaining the fiber material from the plant, for further use.

Preparation of Nanofiber from *B. vulgaris* Fiber

Preparation of nanofiber from *B. vulgaris* fibers was done through acid hydrolysis usage (Saniwan et al., 2012 Kaushik et al., 2010). The *B. vulgaris* fibers were steeped in 10 %w/w HCl with applied ultrasonic agitation, by an ultrasonicator (SB25-12DT, Scientz[®], China), for 2 h and at 60 °C. Final washing of the sample material then followed, before the homogenization of the same for 15 min in a Heidolph[®] DIAX 900 (USA) high shear homogenizer instrument for obtaining *B. vulgaris* nanofibers.

B. vulgaris Nanofiber and *B. vulgaris* Lignin Characterizations Percentage yield analyses

The determinations of *B. vulgaris* nanofiber and *B. vulgaris* lignin yields were obtained from the dry weight of the sample material that was isolated based on the initial dry weight of grinded *B. vulgaris*.

Nanofiber Particle Size Analyses

Investigation of *B. vulgaris* nanofiber particle sizes in solution employed use of dynamic light scattering (DLS) by the Zetasizer Nano-S series instrument from Malven[®], USA, at 173° angle of light scattering, 25°C operating temperature, and 120 sec equilibrating time. For carrying out the particle size analysis, 1 mg of the sample material was dispersed in 10 ml C₂H₅OH before being transferred, using a syringe having 0.22 μ m filter coupled to it, into a polystyrene cuvette. This was then placed into in the analysis stage of the Zetasizer equipment for analyzing the *B. vulgaris* nanofiber particle sizes.

Spectra Analyses

Spectra analyses of the *B. vulgaris* nanofiber and *B. vulgaris* lignin were carried out using the Frontier FT-IR, Perkin Elma[®] (UK) machine, by KBr disc, with spectra range between 4000 and 400 cm⁻¹. This was used for assessing the changes in the chemical structure of the polymer, which could have taken place due to modification.

Thermal Behavior Analyses

The *B. vulgaris* nanofiber and *B. vulgaris* lignin thermal behaviors were studied using the STA 6000 model of Thermogravimetric analyzer (from Perkin Elmer[®]).

Morphological Analyses

Morphological investigations of the *B. vulgaris* nanofiber were done via use of the HRSEM model of Scanning Electron Microscopy (SEM) instrument (from Zeiss Auriga[®]) at a voltage of 15 kV. The samples for the SEM analyses were mounted on the stub having two-sided adhesive tape having a thin layer of gold for its coating (Okeniyi et al., 2018; Okeniyi et al., 2017a; Okeniyi et al., 2017b), for the non-conducting *B. vulgaris* nanofibers being investigated. A 350 times the original size were then used as the magnification for taking the images by the SEM equipment, as detailed in Kampeerapappun et al. (2007).

Transmission Electron Microscopy (TEM)

High resolution Transmission electron microscopy (HRTEM) measurements employed the use of the Zeiss Auriga[®] HRTEM instrument at 100kV. Samples of *B. vulgaris* nanofiber were diluted in distilled water at the ratio of 1 part *B. vulgaris* nanofiber to 20 parts distilled water. Copper grids, at 300-mesh grid sizes, were then dipped in the diluted *B. vulgaris* nanofiber solutions before the grids are dried at ambient temperature. The TEM images of the *B. vulgaris* nanofiber samples were then taken with the microscope. For these, high magnifications at 30,000 and 85,000 times the original sample size were then employed for taking images of the *B. vulgaris* nanofiber samples (Ochigbo et al., 2012; Kampeerapappun et al., 2007).

4. Discussion

Percentage Yields of B. vulgaris Nanofiber and Lignin

The images of *B. vulgaris* nanofiber and *B. vulgaris* lignin are presented in Figure 1. The respective percentage yields of these materials are 33.6% and 21.91%, respectively.



Figure 1. Post-processing images of (a) *B. vulgaris* lignin, and (b) *B. vulgaris* nanofiber.

B. vulgaris Nanofiber Size Distribution

The particle size distribution, of the *B. vulgaris* nanofiber sample studied, is presented in Figure 2.



Figure 2. B. vulgaris nanofiber particle size distribution.

From Figure 2, the particle size distribution showed that *B. vulgaris* nanofiber had two granulometric distributions with the peak of the first at 30 nm, while the peak of the second is at 70 nm. Also, it could be observed that range of the prepared nanofiber sizes is from 20 nm to 100 nm, which indicates the fibers prepared in the present study are in the nano-sized scale.

By this, therefore, the fibers could be suitable for use as fillers in the production of nanocomposite films.

Micrographs of the Prepared B. vulgaris Nanofiber

The micrographs, obtained from the SEM and TEM instruments, of the prepared B. vulgaris nanofiber, are shown in Figure 3. From Figure 3a, the SEM depicted the prepared *B. vulgaris* nanofibers as rod-like, whereas the micrograph from the TEM instrument, Figure 3b, indicated the prepared nanofibers ranged from 20 nm to 100 nm in size. These results from the TEM equipment confirm the B. vulgaris particle size distribution from Figure 2 that had been obtained using another instrument, i.e. the Zetasizer Nano-S series. From Figure 3b, it could also be deduced that the amorphous regions of the nanofiber were transversely cleaved by exposure to hydrochloric acid hydrolysis, thereby reducing the fibers sizes from microns to nanometers (Yong et al., 2012; Azazi-Samir et al., 2005). The result obtained herein exhibited agreement that from Liu et al. (2010), from which the size ranging between 50 nm and 100 nm cellulose nanocrystals were produced from the treatment of strands from bamboo by HNO₃-KClO₃ that was followed by acid hydrolysis. In another related study, Krishnan & Ramesh (2013) also reported particle width range from 30 nm to 90 nm for nanofibers obtained from coconut coir fibers, which is just as Zhang et al. (2007) obtained spherically shaped cellulose nanoparticle materials. The difference in shape of the various nanofibers obtained could be ascribed to the type of treatment used. By the morphological results in the present case, the B. vulgaris nanofibers, so prepared, are found appropriate as fillers in the productions of starch nanocomposite films.



Figure 3. Micrographs of the prepared *B. vulgaris* nanofiber (a) SEM (b) TEM.

FTIR Spectra of B. vulgaris Lignin and B. vulgaris Nanofiber

The FTIR spectra from the *B. vulgaris* lignin and the *B. vulgaris* nanofiber are presented in Figure 4. The functional groups distribution proposed for assignments at adsorbed frequencies resulting from the FTIR spectra are presented in Table 1 (Okeniyi et al., 2019; Okeniyi & Popoola, 2017; Okeniyi et al., 2017c; Okeniyi et al., 2016; Coates, 2000).



Figure 4. FTIR Spectra of (a) *B. vulgaris* lignin and (b) *B. vulgaris* nanofiber.

Table 1.	Assignments of FTIR Spectra Frequencies from B.
	vulgaris lignin and <i>B. vulgaris</i> nanofiber.

B. vulgaris	B. vulgaris	Chemical Groups Assignment
Lignin	Nanofiber	
Frequency	Frequency	
(cm ⁻¹)	(cm ⁻¹)	
3364	3335	aliphatic and aromatic O–H stretch
2922,	2899	saturated aliphatic C–H stretch
2167		
1648	1639	carbonyl stretch (conjugated
		ketone)
1597,	-	Aromatic C=C–C rings stretching
1506,		and bending vibrations
1461		
1422	1428	C–H in-plane bend deformation
-	1368	O-H bend induced by groups of
		phenol
1328	1316	C–O breathing of Syringyl ring
1242	-	C–O breathing of Guaiacyl ring
1159	1160, 1105	C–O stretch vibration of alkyl/cyclic ethers
896 834	897	C-H ring deformation from di-
	,	substituted aryl groups
-	709, 666	O–H out-of-plane bending
		vibration

From the presented results in Table 1, the following characteristic absorption bands derived from the FTIR spectrum of *B. vulgaris* lignin include the absorption band at 3364 cm⁻¹ that is attributed to the stretching vibration of O–H in aliphatic OH and aromatic groups, while the bands in the proximities of 2922 cm⁻¹ and of 2167 cm⁻¹ respectively corresponds to saturated aliphatic CH₂'s asymmetrical and symmetrical stretching bands. The 1648 cm⁻¹ absorption band is allocated to the stretching of conjugated

carbonyl groups. Absorption bands at 1422 cm⁻¹, 1506 cm⁻¹ and 1597 cm⁻¹ are attributed to the skeleton vibration of aromatic groups in the *B. vulgaris* lignin. The 1461 cm⁻¹ absorption frequency illustrates aromatic ring vibration and C–H deformations, while the adsorption at 1242 cm⁻¹ frequency follows from guaiacyl (G) ring breathing with C–O.

In the *B. vulgaris* nanofiber, the 1368 cm⁻¹ absorption frequency corresponds to O–H groups of free phenolic compounds. The absorptions at 1316 cm⁻¹ from the *B. vulgaris* nanofiber spectra and at 1328 cm⁻¹ by the *B. vulgaris* lignin spectra indicate syringyl (S) ring breathing with C-O. The adsorbed frequencies at 896 cm⁻¹ and 1159 cm⁻¹ are respectively attributed to C–O deformation and ether stretching. Finally, the adsorption at 834 cm⁻¹ corresponds to ring vibrations and C–H deformation.

The distributions of functional groups for *B. vulgaris* lignin, in this study, conform to the result reported in literature for Kraft and Klason lignin extracted from pine and wood (Faix, 1991; Zheng-Jun et al., 2012b; Ghatak, 2008; Li, 2011; Ibrahim et al., 2006). In more specific terms, in the system of lignin infrared spectra classified by Faix (1991), the extracted *B. vulgaris* lignin was typically of the G-S type of lignin owing to the consideration that the absorption at 1463 cm⁻¹ is of a lower intensity than the band at 1508 cm⁻¹, in that study. Also, the peak of absorption frequency obtained in that study at 1248 cm⁻¹ is of a stronger intensity than the peak that was obtained at 1325 cm⁻¹, by Faix (1991).

The spectrum of FTIR obtained from *B. vulgaris* nanofiber compares well with that of *B. vulgaris* lignin, with the contrast that the absorption bands at 1506 cm⁻¹ and 1597 cm⁻¹, which correspond to aryl ring of C=C–C stretching from the lignin, are absent. The non-availability of these peaks in *B. vulgaris* nanofiber is attributed to lignin elimination as a result of further acid hydrolysis in *B. vulgaris* nanofiber preparation process. Also, the peak at 1242 cm⁻¹ is also absent in the *B. vulgaris* nanofiber, which illustrates the effective elimination of hemicelluloses, lignin and pectin in the *process* of the *B. vulgaris* nanofiber preparation.

Thermal Behaviors of *B. vulgaris* Lignin and *B. vulgaris* Nanofiber

Presented in Figure 5 are the thermal behaviors of *B. vulgaris* lignin and *B. vulgaris* nanofiber, while the essential parameters of degradation data from the thermal characterizations are plotted in Figure 6.





Figure 5. TGA thermograms: (a) *B. vulgaris* lignin; (b) *B. vulgaris* nanofiber.





From Figure 5, it could be noted that three degradation stages occurred in lignin. The evolution of moisture content of the lignin sample was indicated by the initial weight loss that occurred at the temperature of about 96 °C. The evolution of products that are of low molecular weight such as CO₂, CO and CH₄ occurred at 154 °C and this is termed onset degradation (Visakh et al., 2012a; LeVan, 1989). The peak degradation and end degradation occurred at 295 °C and 488 °C respectively with solid residue of about 29.45%. At temperatures above 500 °C, denoting the third stage, weight loss was no more obvious due to the condensation reactions of aromatic rings that often occur concurrently with decomposition reactions, at this stage, (Zheng-Jun et al., 2012; Visakh et al., 2012a; LeVan, 1989). These observed thermal behaviors by B. vulgaris lignin in the present study exhibit similarity with that obtained from reported research works, carried out on other biological materials, by other authors including Shi et al. (2012), Sumin et al. (2012), and Li (2011).

Alternatively, the TGA plotting of the B. vulgaris nanofiber sample occurred in two degradation stages, during the pyrolytic degradation process. From this, an onset degradation temperature occurred at around 200 °C, which was higher than the onset degradation temperature of lignin. In the $2^{\mbox{\scriptsize nd}}$ degradation stage, a sharp decomposition was observed with the peak temperature of decomposition and of end degradation occurring at 328 °C and 500 °C, respectively. Severe weight loss also occurred such that only about 12.2% remained from the initial sample of B. vulgaris nanofiber. The lower percentage of residue that was observed in the B. vulgaris nanofiber, in comparison to that of the B. vulgaris lignin, could be ascribed to the non-availability of mineral substances, generally oxides, in the B. vulgaris nanofiber. The thermogravimetric degradation behavior, detailed in the present work from B. vulgaris nanofiber, exhibits similarity to results obtained by Visakh et al. (2012b) and LeVan (1989).

Based on the foregoing consideration, it could be inferred that *B. vulgaris* lignin exhibits more thermal stability than the *B. vulgaris* nanofiber, under the condition at which the measurements in the study were carried out. This is in spite of the fact that the degradation occurring from *B. vulgaris* lignin began at lower temperature than the degradation from the *B. vulgaris* nanofiber.

5. Conclusion

Lignin and nanofibers were successfully extracted and prepared from *B. vulgaris* (bamboo). The morphology of the prepared *B. vulgaris* nanofibers were revealed, by SEM and TEM analyses, to be rod-shaped having sizes that range from 20 to 100 nm. FTIR showed that the lignin extracted from *B. vulgaris* is G-S lignin type while the *B. vulgaris* nanofibers are mostly devoid of lignin. TGA showed that the *B. vulgaris* lignin was more thermally stable than that of the *B. vulgaris* nanofiber under the condition that they were measured.

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